Fatima Al-Omran* and Adel Abou El-Khair

Chemistry Department, Faculty of Science, Kuwait University
P.O. Box 5969, 13060 Safat, Kuwait

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#### Abstract

The utility of both 3-(1-benzotriazolyl) chalcone derivatives 3a-c and 2-(1-benzotriazolyl)-1,4-pentadien-3-one (18) in the synthesis of some new $2-(1 H)$-pyridone, pyridine, pyrazole and isoxazole derivatives is reported. Antimicrobial and antifungal screening of some selected examples from the synthesized products were carried out. The structure of the newly synthesized compounds was elucidated by elemental analysis, ir, ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C} \mathrm{nmr}$ investigations.


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The increasing importance of both chalcone and benzotriazole derivatives as intermediates to biologically active compounds [1-7] and in continuation of our current interest in the synthesis of polysubstituted heterocycles incorporating benzotraizole moiety as potential pharmaceuticals [8-11], we reported here on the utility of unreported 3-(1benzotriazolyl) chalcone derivatives 3a-c and 2-(1-benzo-triazolyl)-1,4-pentadien-3-one derivative $\mathbf{1 8}$ as building blocks for synthesis of new polyfunctionally substituted heteroaromatic compounds of expected potential antimicrobial and antifungal activities.
Thus, refluxing an equimolar amount of 1-(1-benzotriazolyl)acetone $\mathbf{1}$ with aromatic aldehydes 2a-c in dimethylformamide (DMF) and in the presence of a catalytic amount of triethylamine afforded in each case of 3-(1-benzotriazolyl) chalcones 3a-c in good yield. The reactivity of the new chalcones 3a-c obtained was investigated toward carbon nucleophiles. Treatment of compounds $\mathbf{3 a}-\mathbf{c}$ with malononitrile in ethanolic piperidine solution at refluxed temperature (Method A), furnished in each case, one isolable product that could be formulated as $2-(1 H)$-pyridone $\mathbf{4}$ or its isomer 5 . The compound $\mathbf{5}$ was obtained by an independent synthetic route via treatment of $\mathbf{1}$ with malononitrile in refluxing ethanol to afforded product $\mathbf{6}$. The latter compound was treated with $2 \mathbf{c}$ to afford a deep brown product identified as 5-(1-benzotriazolyl)-4-methyl-6-(p-nitrophenyl)-2-oxo-1,2-dihydropyridine-3carbonitrile (5) with different melting point than the product obtained from reaction of $\mathbf{3 c}$ with malononitrile but with similar spectral data. The compounds $\mathbf{4 c}$-d could be obtained in situ, via a one step process by treatment of $\mathbf{1}$ with p-methoxybenzaldehyde ( $\mathbf{2 b}$ ) or $p$-nitrobenzaldehyde (2c) in refluxing DMF followed by treatment of the reaction mixture with malononitrile (Method B) to afford a product identical in all respects ( mp and spectra) with that obtained from the reaction of compounds $\mathbf{3 b}-\mathbf{c}$ with malononitrile. The formation of $\mathbf{4}$ was assumed to proceed via initial addition of active methylene reagent across the
double bond in $\mathbf{3 c}$ followed by hydrolysis of one of the cyano groups and then cyclization to 4 .
In a similar manner, treatment of chalcones $\mathbf{3 a}, \mathbf{b}$ with malononitrile in ethanolic sodium ethoxide and in the presence of a catalytic amount of piperidine at refluxed temperature gave a product that could be formulated as 7 or its isomer 8. Spectral data seemed to be little help in discriminating 7 or $\mathbf{8}$. However, the structure $\mathbf{8}$ was ruled out on the basis of the reaction 6 with $p$-methoxybenzaldehyde (2b) in ethanolic sodium ethoxide at refluxed temperature to afford a product with different melting point than the product obtained previously from the reaction of $\mathbf{3 b}$ with malononitrile in ethanolic sodium ethoxide. Compound 7 was assumed to proceed via initial addition of the carbanion of malononitrile across the activated double bond system in $\mathbf{3 b}$ followed by the addition of ethoxide ion to one of the cyano groups to the iminoether, which readily undergoes cyclization via a nucleophilic attack of an NH group on a carbonyl carbon. This sequence of events has been recently suggested to account the formation of alkoxy pyridine [10,12].

The reactivity of chalcone $\mathbf{3 b}$ towards nitrogen nucleophile was also investigated. Thus, treatment of compound 3b with hydrazine hydrate in refluxing ethanol afforded yellow crystals that were identified as a pyrazolylbenzotriazole (Scheme 1). The formation of $\mathbf{9}$ is assumed to proceed via condensation of hydrazine with the carbonyl group in chalcone and subsequent cyclisation into the pyrazole derivative 9. This is similar to the well accepted mechanism of the reaction of chalcones with nitrogen nucleophile [13]. Treatment of 1-(1-benzotriazolyl)-acetone $\mathbf{1}$ with p-methoxybenzaldehyde ( $\mathbf{2} \mathbf{b}$ ) followed by treatment of the reaction mixture with benzoylacetonitrile in refluxing DMF and piperidine afforded $\mathbf{1 0}$. The structure of $\mathbf{1 0}$ was established on the basis of its elemental analysis and spectral data. Thus, the ir spectrum of the reaction product shows the NH and two carbonyl functions (ketone and amide) at $v_{\text {max }} 3652,1692$ and $1650 \mathrm{~cm}^{-1}$, respectively (Scheme 1).


2a-c-4a-c: $\mathrm{Ar}=\mathrm{a}: \mathrm{C}_{6} \mathrm{H}_{5} ;$ b: $p-\mathrm{MeOC}_{6} \mathrm{H}_{4}-;$ c: $p-\mathrm{O}_{2} \mathrm{NC}_{6} \mathrm{H}_{4}-$; $\mathrm{Bt}=1$-Benzotriazolyl

Treatment of $\mathbf{8}$ with each of benzylidenemalononitrile or by refluxing it in a mixture of DMF/ethanol in the presence of elemental sulfur and a catalytic amount of piperidine gave compounds $\mathbf{1 1}$ and 12, respectively in good yield [14-16]. Treatment of $\mathbf{1 2}$ with dimethyl acetylenedicarboxylate (DMAD) afforded the thiepine derivative $\mathbf{1 5}$, which is believed to be formed via the non-isolated intermediates $\mathbf{1 3}$ and $\mathbf{1 4}$ (cf. Scheme 2). Similar formation of theipins has been reported from our laboratorie [15-17] as well as by Dopp et al [18]. The structure of the isolated product $\mathbf{1 5}$ was confirmed on the basis of elemental analysis and spectral data. On the other hand, treatment of compound $\mathbf{1 2}$ with enaminone 16 in refluxing dioxane and in the presence of a catalytic amount of acetic acid afforded only the C-1 alkylated product $\mathbf{1 7}$ in a good yield. Both the ir and ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectra indicated that the amino function was not involved in the reaction. The ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectrum indicated the presence of two trans vinylic protons at $\delta_{\mathrm{H}}=5.85$ and 6.39 ppm with $\mathrm{J}=15 \mathrm{~Hz}$. Thus, structure $\mathbf{1 7}$ was established as the reaction product, which was similar to the well accepted reaction of condensed thiophenes with enaminones, reported from our laboratories [16,19] (cf. Scheme 2).

Furthermore, treatment of 1-(1-benzotriazolyl)acetone 1 with an excess of $p$-nitrophenylbenzaldehyde (2c) in DMF and in presence of a catalytic amount of triethylamine at reflux temperature afforded the dienone $\mathbf{1 8}$ in
excellent yield. The products structure was established on the basis of its elemental analysis and spectral data (Scheme 3). The reactivity of compound $\mathbf{1 8}$ towards some nitrogen and carbon nucleophilies was also investigated.

Thus, treatment of $\mathbf{1 8}$ with hydrazine hydrate or hydroxyl amine in ethanol afforded in each case a product that formulated as pyrazolylbenzotriazole and isoxazolylbenzotriazole derivatives $\mathbf{1 9}$ and $\mathbf{2 0}$, respectively. The formation of $\mathbf{1 9}$ and $\mathbf{2 0}$ is assumed to proceed via initial addition of hydrazine or hydroxylamine to the carbonyl group in system 18 and the resulting product then cyclises into the pyrazole or isoxazole derivatives $\mathbf{1 9}$ and $\mathbf{2 0}$ respectively [13].

On other hand, treatment of $\mathbf{1 8}$ with malononitrile or benzoylacetonitrile in refluxing ethanolic piperidine afforded the 2-(1H)-pyridinone derivatives 21 and 22, respectively in good yield. Both elemental analysis and spectral data were in complete agreement with the assigned structures. The structure $\mathbf{2 1}$ and $\mathbf{2 2}$ are considered most likely for the products based on its similarity to the well established behaviour, which was assumed to proceed via initial addition of active methylene reagent across the activated double bond system in dienone $\mathbf{1 8}$ followed by hydrolysis and cyclization via loss of water a molecule [12].
In a similar manner, dienone $\mathbf{1 8}$ reacted with malononitrile in refluxing ethanolic sodium ethoxide to give the
product that was established as 2-ethoxypyridine 3-carbonitrile derivative 23 in good yield. The ir spectrum of compound 23 showed a characteristic absorption band at $2191 \mathrm{~cm}^{-1}$ due to the nitrile function. The structure may be formed via initial addition of the carbanion of malononitrile across the activated double bond system in dienone 18 followed by addition of ethoxide ion to one of the cyano groups which subsequently cyclises via loss of water to afford 2-ethoxypyridine-3-carbonitrile derivative $\mathbf{2 3}$ (Scheme 3) [12].

Scheme 3


Table 1
In vitro Bactericidal and Fungicidal Activity of Newly Synthesized Compounds

| Compound | E-coli | B-subtilis | S-aureus | A-niger | F-oxysporium |
| :--- | :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |  |
| $\mathbf{3 c}$ | ++++ | ++ | +++ | +++ | +++ |
| $\mathbf{7 b}$ | ++++ | +++ | ++++ | +++ | +++ |
| $\mathbf{1 0}$ | ++ | - | +++ | +++ | +++ |
| $\mathbf{1 1}$ | +++ | ++ | ++++ | +++ | +++ |
| $\mathbf{1 2}$ | ++++ | +++ | +++ | +++ | +++ |
| $\mathbf{1 5}$ | ++++ | +++ | ++++ | +++ | +++ |
| $\mathbf{1 7}$ | +++ | +++ | +++ | +++ | +++ |
| $\mathbf{1 8}$ | +++ | ++ | +++ | - | +++ |
| $\mathbf{2 0}$ | +++ | ++ | ++ | - | +++ |
| $\mathbf{2 2}$ | +++ | - | - | ++ | +++ |

[^0]
## Biological Activity.

The biological activities of some newly synthesized compounds were screened for their antifungal activity against Aspergillus niger and Fusarium oxysporium, while the antibacterial activity was tested against Eschirichia coli, Bacillus subtilis and Staphylococcus aureus. Most of the tested sample showed strong antibacterial and fungicidal activity (Table 1). Bacteria and Fungi were maintained on nutrient agar slops (NA) and sabouraud agar (SA), respectively. When calculated, a loop full of baceria was grown on tryplic Saaybroth, while Fungi were subcultured on a yeast nitrogen base supplemented with glucose (YNBG). All media used were of dificagrade.

## EXPERIMENTAL

All melting points are uncorrected; ir spectra ( $\mathrm{KBr} \mathrm{)} \mathrm{were}$ recorded on a Perkin Elmer 2000 FT-ir spectrophotometer. Both ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C} \mathrm{nmr}$ spectra were recorded on a Brucker 400 MHz spectrometer with DMSO- $\mathrm{d}_{6}$ or $\mathrm{CDCl}_{3}$ as solvent using TMS as internal standard with chemical shifts reported in ppm ( $\delta$ ). Mass spectra were measured on GS/MS INCOL XL, Finingan MAT. Microanalyses were performed on a LECO CHNS 932 analyzer.

General Procedure for the Synthesis of 3a-c.
A mixture of $\mathbf{1}(1.75 \mathrm{~g}, 0.01 \mathrm{~mol})$ in DMF $(20 \mathrm{ml})$ containing 2-3 drops of triethylamine was treated with each of arylaldehydes 2a-c ( 0.01 mol ). The reaction mixture was refluxed for $10-12$ hours. The solid product, so formed, was collected by filtration and recrystallized from the proper solvent.

3-(1'-Benzotriazolyl)-4-phenyl-3-buten-2-one (3a).
This compound was recrystallized from ethanol to give yellow crystals in $69 \%$ yield, $\mathrm{mp} 88-89^{\circ} \mathrm{C}$; ir: $v_{\text {max }} 1697 \mathrm{~cm}^{-1}(\mathrm{CO})$, ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}}: 2.68$ (s, 3H, Me); 6.45 (s, 1H, =CH); 7.28-8.45 ppm (m, 9H, Ar-H); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): 194.78 (CO), 163.27, 145.80, 143.11, 134.09, 132.29, 130.93, 129.94, $128.58,125.38,120.33,118.81$ and 111.28 (aromatic and vinylic carbons) and 26.63 ppm (Me).

Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}: \mathrm{C}, 72.99$; $\mathrm{H}, 4.97$; $\mathrm{N}, 15.95$. Found: C, 72.79; H, 5.06; N, 15.79.

3-(1'-Benzotriazolyl)-4-(p-methoxyphenyl)-3-buten-2-one (3b).
This compound was recrystallized from ethanol to give yellow crystals in $81 \%$ yield, $\mathrm{mp} 123-125^{\circ} \mathrm{C}$, ir: $v_{\max }: 1675 \mathrm{~cm}^{-1}(\mathrm{CO})$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): $\delta_{\mathrm{H}}: 2.72$ (s, 3H, Me); 3.82 (s, 3H, OMe); $6.38(\mathrm{~s}, 1 \mathrm{H},=\mathrm{CH}) ; 6.73-8.38 \mathrm{ppm}(\mathrm{m}, 8 \mathrm{H}, \mathrm{Ar}-\mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): $\delta_{\mathrm{C}} 194.49$ (CO), 162.77, 159.72, 145.91, 142.92, 134.07, 133.38, 129.94, 125.31, 124.34, 120.38, 115.81 and 111.27 (aromatic and vinylic carbons), 56.27 ( OMe ) and 26.63 ppm (Me).

Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2}$ : C, $69.61 ; \mathrm{H}, 5.15 ; \mathrm{N}, 14.32$. Found : C, 69.79; H, 5.08; N, 14.47.

3-(1'-Benzotriazolyl)-4-(p-nitrophenyl)-3-buten-2-one (3c).
This compound was recrystallized from a mixture of ethanol/DMF (2:1) as brown crystals in 71\% yield; mp 197-199 ${ }^{\circ} \mathrm{C}$; ir: $v_{\max } 1662(\mathrm{CO}) ;{ }^{1} \mathrm{H} \operatorname{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}} 2.67$ (s, 3H,

Me); 7.13-8.50 ppm (m, 9H, Ar-H \& =CH); ms (EI): m/z= 308 $\left(\mathrm{M}^{+}\right)$.
Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 62.33; $\mathrm{H}, 3.92 ; \mathrm{N}, 18.17$. Found: C, 62.24; H, 3.86; N, 18.12.
General Procedure for the Synthesis of 4a-c.
Method A.
A mixture of compound 3a-c ( 0.01 mol ) and malononitrile $(0.66 \mathrm{~g}, 0.01 \mathrm{~mol})$ in absolute ethanol $(20 \mathrm{ml})$ and a few drops of piperidine was refluxed for 4 hours, then poured onto cold water. The solid product, so formed, was collected by filtration and recrystallized from ethanol as pale yellow crystals.
Method B.
A mixture of compound $1(1.75 \mathrm{~g}, 0.01 \mathrm{~mol})$ and $p$-methoxybenzaldehyde ( $\mathbf{2 b}$ ) ( $1.36 \mathrm{~g}, 0.1 \mathrm{~mol}$ ) or $p$-nitrobenzaldehyde ( $\mathbf{2 c}$ ) $(1.51 \mathrm{~g}, 0.01 \mathrm{~mol})$ in DMF $(20 \mathrm{ml})$ and a few drops of piperidiene was refluxed for 8 hours. The mixture was then treated with malononitrile ( $0.66 \mathrm{~g}, 0.1 \mathrm{~mol}$ ), refluxed for 5 hours, then allowed to cool, and poured onto ice-cold water. Each of the solid product so formed, was collected by filtration and recrystallized from a mixture of EtOH/DMF (2:1).

5-(1'-Benzotriazolyl)-6-methyl-2-oxo-4-phenyl-1,2-dihydropyri-dine-3-carbonitrile (4a).
This compound was obtained as pale yellow crystals in $75 \%$ yield, mp 175-177 ${ }^{\circ} \mathrm{C}$, ir: $v_{\text {max }}: 3411(\mathrm{NH}), 2193(\mathrm{CN})$ and 1646 $\mathrm{cm}^{-1}(\mathrm{CO}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{6}\right): \delta_{\mathrm{H}} 2.20$ (s, 3H, Me), 7.03-7.65 (m, $9 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ) and 10.20 ppm (br, $1 \mathrm{H}, \mathrm{NH}, \mathrm{D}_{2} \mathrm{O}$-exchangeable).
Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}: \mathrm{C}, 69.71 ; \mathrm{H}, 4.00 ; \mathrm{N}, 21.40$. Found: C, 70.03; H, 4.08; N, 21.31.
5-(1'-Benzotriazolyl)-4-(p-methoxyphenyl)-6-methyl-2-oxo-1,2-dihydropyridine-3-carbonitrile (4b).
This compound was obtained as deep brown crystals in $85 \%$ yield, $\mathrm{mp} 107-109{ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }}: 3336(\mathrm{NH}), 2186(\mathrm{CN})$ and 1658 $\mathrm{cm}^{-1}(\mathrm{CO}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} 2.31(\mathrm{~s}, 3 \mathrm{H}, \mathrm{Me}), 3.81(\mathrm{~s}, 3 \mathrm{H}$, OMe), 6.81-8.21 (m, 8H, Ar-H), $9.50 \mathrm{ppm}\left(\mathrm{b}, 1 \mathrm{H}, \mathrm{NH}, \mathrm{D}_{2} \mathrm{O}-\right.$ exchangeable); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{C}} 163.16$ (C-2), 161.77 (C-6), 158.33, 146.05, 133.98, 131.41, 130.30, 129.52, 128.62, $128.15,125.15,120.14,119.07,115.10,114.17$ (aromatic carbons \& CN ), $111.50(\mathrm{C}-3), 55.51(\mathrm{OMe})$ and $25.45 \mathrm{ppm}(\mathrm{Me})$.
Anal. Calcd. For $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{2}$ : C, 67.22, H, 4.23, $\mathrm{N}, 19.59$. Found: C, 67.36, H, 4.18; N, 19.46.

5-(1'-Benzotriazolyl)-6-methyl-4-(p-nitrophenyl)-2-oxo-1,2-dihydropyridine-3-carbonitrile (4c).

This compound was obtained as deep brown crystals in $92 \%$ yield, mp 160-162 ${ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }}: 3345(\mathrm{NH}), 2203(\mathrm{CN})$ and 1631 $\mathrm{cm}^{-1}(\mathrm{CO}) ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO- $\mathrm{d}_{6}$ ): $\delta_{\mathrm{H}} 2.21$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{Me}$ ), 7.18-8.26 (m, $8 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 10.60 \mathrm{ppm}(\mathrm{b}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{C}} 166.77$ (C-2); 165.12 (C-6), 154.00, 149.72, 145.94, 143.70, 136.12, $133.90,131.41,128.76,127.53,125.10,124.73,120.18$ (aromatic carbons), $118.21(\mathrm{CN}), 115.58$ (C-3) and $21.83 \mathrm{ppm}(\mathrm{Me})$.
Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{12} \mathrm{~N}_{6} \mathrm{O}_{3}$ : C, $61.29 ; \mathrm{H}, 3.24$; N , 22.56.Found: C, 61.39 ; H, 3.24; N, 22.65 .

5-(1'-Benzotriazolyl)-4-methyl-6-(p-nitrophenyl)-2-oxo-1,2-dihydropyridine-3-carbonitrile (5).

A mixture of compound $6(2.41 \mathrm{~g}, 0.01 \mathrm{~mol})$ and $p$-nitrobenzaldehyde $2 \mathbf{c}(1.51 \mathrm{~g}, 0.01 \mathrm{~mol})$ in absolute ethanol $(20 \mathrm{ml})$ was
refluxed for 5 hours, then allowed to cool, poured onto ice-cold water. The solid product, so formed, was collected by filtration and recrystallized from a mixture of EtOH/DMF (2:1) as deep brown crystals in $71 \%$ yield, mp $110-112^{\circ} \mathrm{C}$. ir: $v_{\text {max }}$ : $3345(\mathrm{NH}), 2203$ $(\mathrm{CN})$ and $1631 \mathrm{~cm}^{-1}(\mathrm{CO}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}} 2.31(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{Me}) ; 7.18$-8.26, (m, 8H, Ar-H); $10.50 \mathrm{ppm}(\mathrm{b}, \mathrm{H}, \mathrm{NH})$.
Anal. Calcd. $\mathrm{C}_{19} \mathrm{H}_{12} \mathrm{~N}_{6} \mathrm{O}_{3}$ : C, 61.29; H, 3.24; N, 22.56. Found: C, 61.46; H, 3.40; N, 22.79.
4-(1'-Benzotriazolyl)-2-cyano-3-methyl-2-butenamide (6).
A mixture of $1(1.75 \mathrm{~g}, 0.01 \mathrm{~mol})$ and malononitrile $(0.66 \mathrm{~g}$, $0.01 \mathrm{mmol})$ in ethanol $(20 \mathrm{ml})$ and few drops of triethylamine. The reaction mixture was refluxed for 2 hours, then poured onto ice-cold water and neutralized with $\mathrm{HCl}(10 \%)$. The solid product so formed was collected by filtration and recrystallized from ethanol to give yellow crystal in $89 \%$ yield; mp $130-132^{\circ} \mathrm{C}$; ir: $v_{\text {max }}: 3405-3339\left(\mathrm{NH}_{2}\right), 2190(\mathrm{CN})$ and $1666 \mathrm{~cm}^{-1}(\mathrm{CO}) ;{ }^{1} \mathrm{H}$ nmr (DMSO-d $\mathrm{d}_{6}$ ); $\delta_{\mathrm{H}} 2.30$ (s, 3H, Me), 6.02 ( $\mathrm{s}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ), 7.27$8.10(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}-\mathrm{H})$ and $8.60 \mathrm{ppm}\left(\mathrm{b}, 2 \mathrm{H}, \mathrm{NH}_{2}\right)$.

Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{~N}_{5} \mathrm{O}$ : C, 59.74; H, 4.60; N, 29.03. Found: C, 59.82; H, 4.70; N, 29.10.

General Procedure for the Synthesis of 7a-b.
A solution of each 3a-b ( 0.01 mol ) in absolute ethanol ( 30 ml ) was treated with malononitrile $(0.66 \mathrm{~g}, 0.01 \mathrm{~mol})$ and sodium ethoxide (prepared from 0.60 g of sodium metal and 60 ml of ethanol). The mixture was refluxed for 5 hours, then poured onto ice cold water and neutralized with hydrochloric acid ( $10 \%$ ). Each of the solid product so formed was collected by filtration and recrystallized from ethanol.
5-(1'-Benzotriazolyl)-2-ethoxy-6-methyl-4-phenylpyridine-3carbonitrile (7a).
This compound was obtained as brown crystals in $76 \%$ yield; mp $135-137^{\circ} \mathrm{C}$; ir: $v_{\text {max }}: 2203 \mathrm{~cm}^{-1}(\mathrm{CN}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta$ 1.06 (t, $3 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{Me}$ ), 2.22 (s, $3 \mathrm{H}, \mathrm{Me}$ ), 3.45 (q, 2H, J=7Hz, $\mathrm{OCH}_{2}$ ) and $7.16-8.32 \mathrm{ppm}(\mathrm{m}, 9 \mathrm{H}, \mathrm{Ar}-\mathrm{H})$.
Anal. Calcd. $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}: \mathrm{C}, 70.96 ; \mathrm{H}, 4.82 ; \mathrm{N}, 19.71$. Found: C, 70.79; H, 4.97; N, 19.18.
5-(1'-Benzotriazolyl)-2-ethoxy-4-( $p$-methoxyphenyl)-6-methyl-pyridine-3-carbonitrile (7b).

This compound was obtained as pale brown crystals in $78 \%$ yield; mp 174-176 ${ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }}: 2216 \mathrm{~cm}^{-1}(\mathrm{CN}),{ }^{1} \mathrm{H} \mathrm{nmr}$ $\left(\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} 1.33(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{Me}), 2.30(\mathrm{~s}, 3 \mathrm{H}, \mathrm{Me}), 3.86(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{OMe}$ ), 4.19 ( $\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{OCH}_{2}$ ), 6.40-8.35 (m, 8H, Ar-H); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{C}} 168.36$ (C-2), 166.32 (C-6), 161.58, 159.24, 134.41, 132.16, 131.09, 130.64, 130.00, 129.72, 129.12, $128.65,119.97,117.45,114.87$ and 111.31 (aromatic carbons and $\mathrm{CN}), 55.86$ and $52.92\left(\mathrm{OCH}_{2} \& \mathrm{OMe}\right), 25.48(\mathrm{Me})$ and 16.08 ppm (Me).

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{2}$ : C, 68.56, H, 4.97, $\mathrm{N}, 18.17$. Found: C, $68.38, \mathrm{H}, 4.86, \mathrm{~N}, 18.36$.

5-(1'-Benzotriazolyl)-2-ethoxy-6-(p-methoxyphenyl)-4-methyl-pyridine-3-carbonitrile hydrochloride (8).

A mixture of compound $\mathbf{6}$ was treated with $p$-methoxybenzaldehyde ( $1.54 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) in sodium ethoxide solution ( 30 ml of ethanol in 0.6 g sodium). The reaction mixture was refluxed for 3 hours, left to cool then acidified with $10 \%$ of HCl . The solid product, so formed, was collected by filtration and recrystallized from a
mixture of $\mathrm{EtOH} / \mathrm{DMF}$ (2:1) to give pale yellow crystals in $92 \%$ yield; mp 140-142 ${ }^{\circ} \mathrm{C}$ ir: $v_{\text {max }}: 2216 \mathrm{~cm}^{-1}(\mathrm{CN}) .{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO$\mathrm{d}_{6}$ ): $\delta_{\mathrm{H}} 1.12(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7 \mathrm{H}), 2.37(\mathrm{~s}, 3 \mathrm{H}, \mathrm{Me}) ; 3.81$ ( $\left.\mathrm{s}, 3 \mathrm{H}, \mathrm{OMe}\right)$; 4.16 (q, 2H, J=7Hz, OCH 2 ); 6.72-7.88 (m, 8H, Ar-H).

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{20} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{Cl}: \mathrm{C}, 62.63 ; \mathrm{H}, 4.77 ; \mathrm{N}, 16.60$. Found: C, 62.50; H, 5.11; N, 16.32.

1-[(3'-Methyl-(1H)-5'-(p-methoxyphenyl)pyrazol-4'-yl)]benzotriazole (9).

A solution of $\mathbf{3 b}(2.93 \mathrm{~g}, 0.01 \mathrm{~mol})$ in ethanol $(20 \mathrm{ml})$ was treated with hydrazine hydrate $(0.5 \mathrm{~g}, 0.01 \mathrm{~mol})$. The reaction mixture was refluxed for 2 hours then allowed to cool. The solid product, so formed, was collected by filtration and recrystallized from ethanol to give yellow crystals in 73\% yield; mp 160-162 ${ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }} 3402 \mathrm{~cm}^{-1}(\mathrm{NH}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{\mathrm{d}}^{6}\right): \delta_{\mathrm{H}} 2.17$ (s, $3 \mathrm{H}, \mathrm{Me}$ ), 3.42 (s, 3H, OMe), 6.20-8.20 (m, 8H, Ar-H) and 10.50 ppm (b, 1H, NH).

Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}: \mathrm{C}, 66.87 ; \mathrm{H}, 4.95 ; \mathrm{N}, 22.93$. Found: C, 66.97; H, 5.02; N, 22.83.

5-(1'-Benzotriazolyl)-3-benzoyl-4-(p-methoxyphenyl)-6-methyl-2-oxo-1,2-dihydropyridine (10).

A solution of compound $\mathbf{1}(1.75 \mathrm{~g}, 0.01 \mathrm{~mol})$ in DMF ( 20 ml ) was treated with $p$-methoxybenzaldehyde ( $\mathbf{2 b}$ ) ( $1.21 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) and a few drops of piperidine. The reaction mixture was refluxed for 8 hours then treated with benzoylacetonitrile ( $1.45 \mathrm{~g}, 0.01$ mol ) and refluxed for another 5 hours then poured onto ice-cold water. The solid product, so formed, was collected by filtration and recrystallized from a mixture of $\mathrm{EtOH} / \mathrm{DMF}$ (2:1) to give yellow crystals in $72 \%$ yield, mp $118-120{ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }}: 3352$ (NH), $1692(\mathrm{CO})$ and $1650 \mathrm{~cm}^{-1}(\mathrm{CO}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right) ; \delta_{\mathrm{H}}$ 2.20 (s, 3H, Me), 3.82 (s, 3H, OMe), 6.88-7.95 (m, 13H, Ar-H) and $8.25 \mathrm{ppm}(\mathrm{b}, 1 \mathrm{H}, \mathrm{NH})$.

Anal. Calcd. For $\mathrm{C}_{26} \mathrm{H}_{20} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 71.55; H, 4.62; N, 12.82. Found: C, 71.87; H, 4.32; N, 13.09.

8-Amino-4-(1'-Benzotriazolyl)-1-ethoxy-3-( $p$-methoxyphenyl)-6-phenyl-6,7-dihydroisoquinoline-7,7-dicarbonitrile hydrochloride (11).

A solution of $\mathbf{8}(4.21 \mathrm{~g}, 0.01 \mathrm{~mol})$ in 20 ml of ethanol and few drops of piperidine was treated with benzylidenemalononitrile $(1.54 \mathrm{~g}, 0.01 \mathrm{~mol})$. The reaction mixture was refluxed for 5 hours and then poured onto cold water. The solid product, so formed, was collected by filtration and recrystallized from a mixture of DMF/EtOH (2:1) as brown crystals in $83 \%$ yield; mp 167-169 ${ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }}$ : 3339-3195 $\left(\mathrm{NH}_{2}\right)$; 2223 and $2189 \mathrm{~cm}^{-1}(2 \mathrm{CN}) ;{ }^{1} \mathrm{H}$ $\mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}} 1.34(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}-7 \mathrm{~Hz}, \mathrm{Me}), 2.55(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6)$, 3.75 (s, 3H, OMe), 4.42 (q, 2H, J=7Hz, OCH ${ }_{2}$ ), 6.60-7.86 (m, $14 \mathrm{H}, \mathrm{Ar}-\mathrm{H} \& \mathrm{H}-5), 7.97 \mathrm{ppm}\left(\mathrm{bs}, 2 \mathrm{H}, \mathrm{NH}_{2}\right.$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO$\left.\mathrm{d}_{6}\right): \delta_{\mathrm{C}} 165.68(\mathrm{C}-1), 162.61,161.63\left(\mathrm{C}_{4} \& \mathrm{C}_{8}\right), 155.22,146.11$, 145.71, 144.65, 143.86, 135.41, 131.52, 130.56, 129.68, 129.41, 129.38, 129.28, 129.17, 128.78, 120.11, 119.82, 115.49, 114.68, 111.74 and 107.30 (aromatic carbons 2 CN ) 56.21 and 56.83 $\left(\mathrm{OCH}_{2} \& \mathrm{OCH}_{3}\right), 26.52(\mathrm{C}-6), 21.87(\mathrm{Me})$ and $18.90 \mathrm{ppm}(\mathrm{C}-7)$.
Anal. Calcd. For $\mathrm{C}_{32} \mathrm{H}_{26} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{Cl}: \mathrm{C}, 66.72, \mathrm{H}, 4.54, \mathrm{~N}, 17.02$. Found: C, 66.80, H, 4.29, N, 17.17.

3-Amino-7-(1'-Benzotriazolyl)-4-ethoxy-6-(p-methoxyphenyl)-thieno[3,4-c]pyridine (12).
A solution of $\mathbf{8}(4.21 \mathrm{~g}, 0.01 \mathrm{~mol})$ in 20 ml of $\mathrm{EtOH} / \mathrm{DMF}$ (2:1) was treated with element al sulfur ( $0.32 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) and
piperidine ( 0.2 ml ). The mixture was refluxed for 4 hours then poured onto water. The solid product, so formed, was collected by filtration and recrystallization from a mixture of EtOH/DMF (2:1) to give green crystals in $76 \%$ yield, mp $155-157{ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }}: 3448,3383 \mathrm{~cm}^{-1}\left(\mathrm{~b}, \mathrm{NH}_{2}\right) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{6}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}} 1.21$ (t, $3 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{Me}) ; 3.81$ (s, $3 \mathrm{H}, \mathrm{OMe}) ; 4.14(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}$, $\mathrm{OCH}_{2}$ ), 6.64-7.86 (m, 9H, Ar-H \& H-1) and 8.54 (bs, $2 \mathrm{H}, \mathrm{NH}_{2}$, $\mathrm{D}_{2} \mathrm{O}$-exchangeable).

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}: \mathrm{C}, 63.30 ; \mathrm{H}, 4.59 ; \mathrm{N}, 16.78$. Found: C, 63.29; H, 4.71; N, 16.72.

Dimethyl-4-amino-9-(1'-Benzotriazolyl)-6-ethoxy-8-( $p$-methoxy-phenyl)thiepino[3,4-c]pyridine-2,3-dicarboxylate (15).

A mixture of $12(4.17 \mathrm{~g}, 0.01 \mathrm{~mol})$ and DMAD $(1.23 \mathrm{~g}, 0.01$ $\mathrm{mol})$ in dioxane ( 20 ml ) and acetic acid ( 2 ml ) was refluxed for 8 hours then poured onto ice cold water. The solid product, so formed, was collected by filtration and recrystallized from a mixture of DMF/EtOH (2:1) as green crystals in $74 \%$ yield, mp $170-172{ }^{\circ} \mathrm{C}$; ir: $v_{\max } 3410,3380\left(\mathrm{NH}_{2}\right)$ and $1735 \mathrm{~cm}^{-1}(2 \mathrm{CO})$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{\mathrm{d}}\right.$ ): $\delta_{\mathrm{H}} 1.17(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7 \mathrm{H}, \mathrm{Me}), 3.70(\mathrm{~s}, 3 \mathrm{H}$, OMe ), 3.75 (s, 3H, OMe), 3.80 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{OMe}$ ), 4.05 ( $\mathrm{q}, 2 \mathrm{H}$, $\left.\mathrm{J}=7 \mathrm{~Hz}, \mathrm{OCH}_{2}\right), 6.99(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-1), 7.05-8.00(\mathrm{~m}, 8 \mathrm{H}, \mathrm{Ar}-\mathrm{H})$ and $8.02 \mathrm{ppm}\left(\mathrm{bs}, 2 \mathrm{H}, \mathrm{NH}_{2}\right) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{C}} 166.66(\mathrm{CO})$, 164.85 (CO), 161.61 (C-6), 155.28 (C-8), 148.03, 146.09, 145.57, 144.63, 142.13, 135.39, 130.75, 129.89, 129.25, 128.88, 128.22, 125.38, 123.48,120.46, 117.46, 115.57, 114.68 (aromatic carbons), $67.38\left(\mathrm{OCH}_{2}\right), 56.30(3 \mathrm{OMe})$ and 19.45 ppm (Me).

Anal. Calcd. for $\mathrm{C}_{28} \mathrm{H}_{25} \mathrm{~N}_{5} \mathrm{O}_{6} \mathrm{~S}: \mathrm{C}, 60.10 ; \mathrm{H}, 4.50 ; \mathrm{N}, 12.52$. Found: C, 60.20; H, 4.41; N, 12.81.
3-[3'-Amino-7'-( $1^{\prime \prime}$-benzotriazolyl)-4'-ethoxy-6'-( $p$-methoxy-phenyl)thieno-[3,4c]pyridin-1-yl]-1-(2-thienyl)-2-propen-1-one (17).

To a solution of $\mathbf{1 2}(4.17 \mathrm{~g}, 0.01 \mathrm{~mol})$ in a mixture of dioxane $(20 \mathrm{ml})$ and acetic acid $(2 \mathrm{ml})$ was treated with $3-(N, N-$ dimethylamino)-1-(2'-thienyl)-2-propen-1-one (16) (1.81 g, $0.01 \mathrm{~mol})$. The reaction mixture was refluxed for 8 hours and poured onto ice cold water. The solid product, so formed, was collected by filtration and recrystallized from a mixture of DMF/EtOH (2:1) to give green crystals in $73 \%$ yield, mp 158$160{ }^{\circ} \mathrm{C}$, ir: $v_{\text {max }}: 3447$ and $3342\left(\mathrm{NH}_{2}\right)$ and $1654 \mathrm{~cm}^{-1}(\mathrm{CO})$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO- $\mathrm{d}_{6}$ ): $\delta_{\mathrm{H}} 1.21(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{Me}), 3.81(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{OMe}), 4.18\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{OCH}_{2}\right), 5.85(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=15 \mathrm{~Hz}$, vinylic-H) 6.39 (d, $1 \mathrm{H}, \mathrm{J}=15 \mathrm{~Hz}$, vinylic-H), 6.38-8.30 ( $\mathrm{m}, 11 \mathrm{H}$, Ar-H) and $8.42 \mathrm{ppm}\left(\mathrm{b}, 2 \mathrm{H}, \mathrm{NH}_{2}\right) ;{ }^{13} \mathrm{C}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{C}} 188.99$ (CO ketone), 165.66, 164.84, 161.27, 155.28, 148.04, 145.70, $144.63,143.83,142.12,135.39,130.75,129.89,129.53$, 128.88, 128.32, 125.46, 123.47, 120.46, 115.70, 115.10, 114.68, $111.23,110.85$ (aromatic \& vinylic carbons), $67.37\left(\mathrm{OCH}_{2}\right)$, $56.24(\mathrm{OMe})$ and $19.46 \mathrm{ppm}(\mathrm{Me})$.

Anal. Calcd. for $\mathrm{C}_{29} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}_{3} \mathrm{~S}_{2}$ : C, 62.92, $\mathrm{H}, 4.19, \mathrm{~N}, 12.65$. Found: C, 63.06, H, 4.37, N, 12.40.

2-(1'-Benzotriazolyl)-1,5-bis(p-nitrophenyl)-1,4-pentadien-3one (18).

A mixture of $\mathbf{1}(1.75 \mathrm{~g}, 0.01 \mathrm{~mol})$ in DMF $(20 \mathrm{ml})$ containing 2-3 drops of triethylamine was treated with $p$-nitrobenzaldehyde (2c) $(3.02 \mathrm{~g}, 0.02 \mathrm{~mol})$. The reaction mixture was refluxed for 12-14 hours then poured onto ice-cold water. The solid product, so formed, was collected by filtration and recrystallized from a
mixture of $\mathrm{EtOH} / \mathrm{DMF}$ (2:1) as brown crystals in $81 \%$ yield, mp . $166-168{ }^{\circ} \mathrm{C}$, ir: $v_{\text {max }}: 1662(\mathrm{CO}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{\mathrm{d}}\right.$ ): $\delta_{\mathrm{H}} 6.84$ (s, $1 \mathrm{H},=\mathrm{CH}), 7.01-8.43(\mathrm{~m}, 14 \mathrm{H}$, Ar-H \& vinylic-H).
Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{5}$ : C, 62.58; H, 3.42; N, 15.87. Found: C, 62.58; H, 3.53; N, 15.62.

General Procedure for the Synthesis of 19-20.
A solution of $\mathbf{1 8}(4.41 \mathrm{~g}, 0.01 \mathrm{~mol})$ in 20 ml of ethanol was treated with hydrazine hydrated $(0.5 \mathrm{~g}, 0.01 \mathrm{~mol})$ or hydroxylamine hydrochloride ( $0.69 \mathrm{~g}, 0.01 \mathrm{~mol})$. The reaction mixture was refluxed for 1 hour and left cool at room temperature. Each of the solid product so formed was collected by filtration and recrystllized from ethanol.
1-\{2'-(p-Nitrophenyl)-1'-[5"-(p-nitrophenyl)-1" $H$-pyrazol-3"-yl\} vinyl\}benzotriazole (19).
This compound was obtained as yellow crystals in $61 \%$ yield; mp 210-212 ${ }^{\circ} \mathrm{C}$; ir: $v_{\text {max }}(\mathrm{NH}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}} 6.30$ (s, 1 H , pyrazol-H), $6.72(\mathrm{~s}, 1 \mathrm{H},=\mathrm{CH}), 7.22-8.22(\mathrm{~m}, 12 \mathrm{H}, \mathrm{Ar}-\mathrm{H})$ and $8.80 \mathrm{ppm}\left(\mathrm{b}, 1 \mathrm{H}, \mathrm{D}_{2} \mathrm{O}\right.$-exchangeable NH$)$.
Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{15} \mathrm{~N}_{7} \mathrm{O}_{4}$ : C, 60.87; H, 3.33; N, 21.62. Found: C, 60.76; H, 3.13; N. 21.82.
1-\{2'-(p-Nitrophenyl)-1'-[5"-(p-nitrophenyl)isoxazol-3"-yl]vinyl $\}$ benzotriazole (20).
This compound was obtained as a yellow brown solid in $79 \%$ yield, mp 190-192 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{\mathrm{d}}\right.$ ): $\delta_{\mathrm{H}} 6.43$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H}-4$ ), $6.82(\mathrm{~s}, 1 \mathrm{H},=\mathrm{CH}), 7.01-8.31(\mathrm{~m}, 12 \mathrm{H}, \mathrm{Ar}-\mathrm{H})$.

Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{14} \mathrm{~N}_{6} \mathrm{O}_{5}$ : C, $60.79 ; \mathrm{H}, 3.10 ; \mathrm{N}, 18.49$. Found: C, 61.03; H, 3.32; N, 18.67.
General Procedure for the Synthesis of 21-22.
To suspension of compound $18(4.41 \mathrm{~g}, 0.01 \mathrm{~mol})$ in 20 ml of ethanol was treated with malononitrile $(0.66 \mathrm{~g}, 0.01 \mathrm{~mol})$ or benzoylacetonitrile $(1.45 \mathrm{~g}, 0.01 \mathrm{~mol})$ and few drops of piperidine. The reaction mixture was refluxed for 6 hours. The solvent was evaporated under reduced pressure. Each of the solid product, so formed, was collected by filtration and recrystallized from proper solvent.

6-[1'-(1-Benzotriazolyl)-2'-(p-nitrophenyl)vinyl]-4(p-nitrophenyl)-2-oxo-1,2-dihydropyridine-3-carbonitrile hydrochloride (21).

This compound was recrystallized from ethanol as brown crystals in $76 \%$ yield; $\mathrm{mp} 180-182^{\circ} \mathrm{C}$; ir: $v_{\text {max }} 3379(\mathrm{~b}, \mathrm{NH}), 2192(\mathrm{CN})$ and $1657 \mathrm{~cm}^{-1}(\mathrm{CO}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}} 6.68(\mathrm{~s}, 1 \mathrm{H},=\mathrm{CH})$, 7.03-8.30 (m, 13H, Ar-H \& H-5) and $10.80 \mathrm{ppm}(\mathrm{b}, 1 \mathrm{H}, \mathrm{NH})$.

Anal. Calcd. for $\mathrm{C}_{26} \mathrm{H}_{16} \mathrm{~N}_{7} \mathrm{O}_{5} \mathrm{Cl}$ : C, $57.62 ; \mathrm{H}, 2.97 ; \mathrm{N}, 18.09$. Found: C, 57.69; H, 3.14; N, 18.20.

6-[1'-(1-Benzotriazolyl)-2'(p-nitrophenyl)vinyl]-3-benzoyl-4-(p-nitrophenyl)-2-oxo-1,2-dihydropyridine-3-carbonitrile (22).

This compound was recrystallized from a mixture of EtOH/DMF (2:1) as reddish brown crystals in 73\% yield; mp 163$165^{\circ} \mathrm{C}$; ir: $v_{\max } 3367(\mathrm{NH}), 1697$ and $1632 \mathrm{~cm}^{-1}$ (2CO); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): $\delta_{\mathrm{H}} 7.01$ (s, 1H, =CH), 7.03 (s, 1H, H-5), 7.29-7.98 (m, 17H, Ar-H), 8.05 ppm (br, $1 \mathrm{H}, \mathrm{NH}, \mathrm{D}_{2} \mathrm{O}$ exchangeable).
Anal. Calcd for $\mathrm{C}_{32} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{O}_{6}$ : C, $65.75 ; \mathrm{H}, 3.45 ; \mathrm{N}, 14.38$. Found: C, 65.38; H, 3.37; N, 14.37.
6-[1'-(1-Benzotriazolyl)-2'-(p-nitrophenyl)vinyl]-2-ethoxy-4-(p-nitrophenyl)pyridine-3-carbonitrile hydrochloride (23).

A mixture of 18 ( $4.41 \mathrm{~g}, 0.01 \mathrm{~mol})$ and malononitrile $(0.66 \mathrm{~g}$, 0.01 mol ) in absolute ethanol ( 50 ml ) was treated with sodium
ethoxide (prepared from 0.6 g sodium metal and 60 ml of ethanol). The reaction mixture was refluxed for 3 hours, then poured onto cold water and neutralized with hydrochloric acid (10\%). The solid product was collected by filtration and recrystallized from a mixture of EtOH/DMF (2:1) as brown crystals $68 \%$ yield, mp $250-252^{\circ} \mathrm{C}$ ir: $v_{\max } 2191 \mathrm{~cm}^{-1}(\mathrm{CN}) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}}$ $1.23(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{Me}), 4.36\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{OCH}_{2}\right), 6.94(\mathrm{~s}, 1 \mathrm{H}$, $=\mathrm{CH}$ ), $7.22-8.13 \mathrm{ppm}(\mathrm{m}, 13 \mathrm{H}, \mathrm{Ar}-\mathrm{H} \& \mathrm{H}-5)$.

Anal. Calcd. for $\mathrm{C}_{28} \mathrm{H}_{20} \mathrm{~N}_{7} \mathrm{O}_{5} \mathrm{Cl}: \mathrm{C}, 59.00 ; \mathrm{H}, 3.53 ; \mathrm{N}, 17.20$. Found: C, 59.12; H, 3.43; N, 17.01.

## Biological Testing.

The newly synthesized compounds were tested against the specified microorganism, using $400 \mu \mathrm{~g} / \mathrm{mL}$ ( $\mathrm{w} / \mathrm{v}$ ) solutions in sterile dimethyl-d ${ }_{6}$ sulfoxide (DMSO). A solution of the tested compound ( 0.01 mol ) was poured aseptically in a well of 6 mm diameter made by a Cork borer in the nutrient agar medium for bacterial test and in Sabourand agar for fungal test. After placing the same volume in wells of all tested microorganisms nutrient agar plates were incubated at $37^{\circ} \mathrm{C}$ for 24 h and sabourand dextrose agar plates were incubated at $25^{\circ} \mathrm{C}$ for 48 hours The activities were expressed as inhibition zones ( mm , diameter, as clear areas) as antibacterial and antifungal effect. The least concentration, which showed inhibitory effect on any specific micoorganism, was considered as the minimum inhibitory concentration (MIC) that was determined using streptomycin $(50 \mu \mathrm{~g} / \mathrm{ml})$ as the references.

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[^0]:    No effect $=-$; slight effect $=+$; Moderate effect $=++$; strong effect $=$ +++, ++++

